# **Lucitone Liquid**

#### **ChemWatch Review SDS**

Chemwatch: 4993-47 Version No: 3.1.1.1

Safety Data Sheet according to WHS and ADG requirements

Chemwatch Hazard Alert Code: 3

Issue Date: **01/01/2013** Print Date: **22/11/2016** S.GHS.AUS.EN

## SECTION 1 IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

# Product Identifier Product name Lucitone Liquid Synonyms ?, Lucitone Liquid Proper shipping name METHYL METHACRYLATE MONOMER, STABILISED

# Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses Denture acrylic resin.

Other means of identification

#### Details of the supplier of the safety data sheet

Registered company name	Dentsply (Australia) Pty Ltd			
Address	11-21 Gilby Road Mount Waverley VIC 3149 Australia			
Telephone	1300 55 29 29			
Fax	+61 3 9538 8260			
Website	www.dentsply.com.au			
Email	clientservices@dentsply.com			

#### Emergency telephone number

Association / Organisation	Poisons Information Centre (AUSTRALIA)	
Emergency telephone numbers	13 11 26 - AUSTRALIA (24 hour service)	
Other emergency telephone numbers	Not Available	

# **SECTION 2 HAZARDS IDENTIFICATION**

#### Classification of the substance or mixture

## HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

#### CHEMWATCH HAZARD RATINGS

	Min	Max	
Flammability	3		
Toxicity	2		0 = Minimum
Body Contact	2		1 = Low 2 = Moderate
Reactivity	2		3 = High
Chronic	2		4 = Extreme

Poisons Schedule	Not Applicable
Classification <sup>[1]</sup>	Flammable Liquid Category 2, Skin Corrosion/Irritation Category 2, Eye Irritation Category 2A, Skin Sensitizer Category 1, Specific target organ toxicity - single exposure Category 3 (respiratory tract irritation), Specific target organ toxicity - single exposure Category 3 (narcotic effects)
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HSIS ; 3. Classification drawn from EC Directive 1272/2008 - Annex VI

#### Label elements

GHS label elements





SIGNAL WORD	DANGE

#### Hazard statement(s)

H225	Highly flammable liquid and vapour.
H315	Causes skin irritation.
H319	Causes serious eye irritation.

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H317	May cause an allergic skin reaction.				
H335	May cause respiratory irritation.				
H336	May cause drowsiness or dizziness.				
AUH019	May form explosive peroxides				
Precautionary statement(s)	) Prevention				
P210	Keep away from heat/sparks/open flames/hot surfaces No smoking.				
P271	Use only outdoors or in a well-ventilated area.				
P280	Wear protective gloves/protective clothing/eye protection/face protection.				
P240	Ground/bond container and receiving equipment.				
Precautionary statement(s)	) Response				
P362	P362 Take off contaminated clothing and wash before reuse.				
P363 Wash contaminated clothing before reuse.					
P370+P378	In case of fire: Use alcohol resistant foam or normal protein foam for extinction.				
P302+P352	IF ON SKIN: Wash with plenty of soap and water.				
Precautionary statement(s)	Precautionary statement(s) Storage				
P403+P235	P403+P235 Store in a well-ventilated place. Keep cool.				
P405	Store locked up.				
Precautionary statement(s)	) Disposal				
P501	Dispose of contents/container in accordance with local regulations.				

# SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

#### Substances

See section below for composition of Mixtures

#### Mixtures

CAS No	%[weight]	Name	
80-62-6	>60	methyl methacrylate	
97-90-5	<10	ethylene glycol dimethacrylate	

## **SECTION 4 FIRST AID MEASURES**

## Description of first aid measures

Eye Contact	If this product comes in contact with the eyes:  Immediately hold eyelids apart and flush the eye continuously with running water.  Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.  Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.  Transport to hospital or doctor without delay.  Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
Skin Contact	If skin contact occurs:  ► Immediately remove all contaminated clothing, including footwear.  ► Flush skin and hair with running water (and soap if available).  ► Seek medical attention in event of irritation.
Inhalation	<ul> <li>If fumes or combustion products are inhaled remove from contaminated area.</li> <li>Lay patient down. Keep warm and rested.</li> <li>Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.</li> <li>Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.</li> <li>Transport to hospital, or doctor, without delay.</li> </ul>
Ingestion	<ul> <li>If swallowed do NOT induce vomiting.</li> <li>If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.</li> <li>Observe the patient carefully.</li> <li>Never give liquid to a person showing signs of being sleepy or with reduced awareness; i.e. becoming unconscious</li> <li>Give water to rinse out mouth, then provide liquid slowly and as much as casualty can comfortably drink.</li> <li>Seek medical advice.</li> </ul>

## Indication of any immediate medical attention and special treatment needed

Treat symptomatically.

#### **SECTION 5 FIREFIGHTING MEASURES**

# Extinguishing media

- ► Foam.
- ► Dry chemical powder.

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- ▶ BCF (where regulations permit).
- Carbon dioxide.

#### Special hazards arising from the substrate or mixture

# Fire Incompatibility

- ▶ Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result
- Polymerisation may occur at elevated temperatures
- Polymerisation may be accompanied by generation of heat as exotherm.
- Process is self accelerating as heating causes more rapid polymerisation.
- Exotherm may cause boiling with generation of acrid, toxic and flammable vapour.
- ▶ Contamination with polymerisation catalysts peroxides, persulfates, oxidising agents also strong acids, strong alkalies, will cause polymerisation with exotherm - generation of heat.
- ▶ Polymerisation of large quantities may be violent even explosive.

#### Advice for firefighters

#### Fire Fighting

- ▶ Alert Fire Brigade and tell them location and nature of hazard.
- May be violently or explosively reactive
- ▶ Wear full body protective clothing with breathing apparatus.
- ▶ Prevent, by any means available, spillage from entering drains or water course.

▶ Hot organic vapours or mist are capable of sudden spontaneous combustion when mixed with air even at temperatures below their published autoignition temperatures

- The temperature of ignition decreases with increasing vapour volume and vapour/air contact times and is influenced by pressure change.
- Ignition may occur under elevated-temperature process conditions especially in processes performed under vacuum subjected to sudden ingress of air or in processes performed at elevated pressure, where sudden escape of vapours or mists to the atmosphere occurs.

#### Fire/Explosion Hazard

Combustion products include: carbon dioxide (CO2)

nitrogen oxides (NOx)

other pyrolysis products typical of burning organic material.

- ▶ Liquid and vapour are highly flammable.
- ► Severe fire hazard when exposed to heat, flame and/or oxidisers.
- Vapour may travel a considerable distance to source of ignition.
- ▶ Heating may cause expansion or decomposition leading to violent rupture of containers

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#### **SECTION 6 ACCIDENTAL RELEASE MEASURES**

#### Personal precautions, protective equipment and emergency procedures

See section 8

#### **Environmental precautions**

See section 12

#### Methods and material for containment and cleaning up

Minor	Spil

► Remove all ignition sources.

Restrict access to area

- Clean up all spills immediately.
- Avoid breathing vapours and contact with skin and eves.
- ▶ Control personal contact with the substance, by using protective equipment.

#### Major Spills

- ► Clear area of personnel and move upwind.
- ▶ Alert Fire Brigade and tell them location and nature of hazard.
- ▶ May be violently or explosively reactive.
- ▶ Wear breathing apparatus plus protective gloves.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

#### **SECTION 7 HANDLING AND STORAGE**

Safe handling

## Precautions for safe handling

- Most acrylic monomers have low viscosity therefore pouring, material transfer and processing of these materials do not necessitate heating.
- Viscous monomers may require heating to facilitate handling. To facilitate product transfer from original containers, product must be heated to no more than 60 deg. C. (140 F.), for not more than 24 hours.
- Avoid all personal contact, including inhalation.
- Wear protective clothing when risk of exposure occurs.
- ► Use in a well-ventilated area.
- Prevent concentration in hollows and sumps.
- ▶ DO NOT allow clothing wet with material to stay in contact with skin

#### Store in original containers in approved flame-proof area. No smoking, naked lights, heat or ignition sources

- DO NOT store in pits, depressions, asements or areas where vapours may be trapped.
- Keep containers securely sealed.
- Storage requires stabilising inhibitor content and dissolved oxygen content to be monitored. Refer to manufacturer's recommended levels.
- DO NOT overfill containers so as to maintain free head space above product
- ▶ Blanketing or sparging with nitrogen or oxygen free gas will deactivate stabiliser.

# Other information

Easily peroxidisable.

- Products formed as a result of peroxidation are not only safety hazards but may chemically alter the chemical behavior of the parent compound.
- For Should have a warning label affixed bearing the date of receipt in the laboratory and the date on which the container label is first opened, or laboratory synthesised materials are the responsibility of the individual chemist.
- ▶ WARNING: This product may form peroxides which themselves are not themselves particularly hazardous but which on decomposition may initiate explosive polymerisation of the bulk monomer (Trommsdorf effect).

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Conditions for safe storage	e, including any incompatibilities
Suitable container	<ul> <li>Packing as supplied by manufacturer.</li> <li>Plastic containers may only be used if approved for flammable liquid.</li> <li>Check that containers are clearly labelled and free from leaks.</li> <li>For low viscosity materials (i): Drums and jerry cans must be of the non-removable head type. (ii): Where a can is to be used as an inner package, the can must have a screwed enclosure.</li> <li>For materials with a viscosity of at least 2680 cSt. (23 deg. C)</li> <li>For manufactured product having a viscosity of at least 250 cSt.</li> </ul>
Storage incompatibility	Segregate from strong acids strong alkalis ammonia amines oxidisers and strong oxidisers  • Contamination with polymerisation catalysts - peroxides, persulfates, oxidising agents - also strong acids, strong alkalies, will cause polymerisation with exotherm - generation of heat.

## **SECTION 8 EXPOSURE CONTROLS / PERSONAL PROTECTION**

▶ Polymerisation of large quantities may be violent - even explosive.

#### **Control parameters**

#### OCCUPATIONAL EXPOSURE LIMITS (OEL)

#### INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
Australia Exposure Standards	methyl methacrylate	Methyl methacrylate	208 mg/m3 / 50 ppm	416 mg/m3 / 100 ppm	Not Available	Not Available

#### **EMERGENCY LIMITS**

Ingredient	Material name	TEEL-1		TEEL-2	TEEL-3
methyl methacrylate	Methyl methacrylate	Not Available		Not Available	Not Available
ethylene glycol dimethacrylate	Ethylene glycol dimethacrylate	9.9 mg/m3		110 mg/m3	650 mg/m3
In the Paris	Octobro LIBITI		n		

Ingredient	Original IDLH	Revised IDLH
methyl methacrylate	4,000 ppm	1,000 ppm
ethylene glycol dimethacrylate	Not Available	Not Available

# **Exposure controls**

#### Appropriate engineering controls

Use in a well-ventilated area

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment

#### Personal protection









# Eye and face protection

- Safety glasses with side shields
- Chemical goggles
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lenses or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience.

#### Skin protection

See Hand protection below

#### Hands/feet protection

- ▶ Wear chemical protective gloves, e.g. PVC.
- ▶ Wear safety footwear or safety gumboots, e.g. Rubber NOTE:

- F The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact.
- ▶ Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.

# **Body protection**

See Other protection below

#### Other protection

- Overalls.
- ► PVC Apron.
  - PVC protective suit may be required if exposure severe.
  - ► Eyewash unit.

#### Thermal hazards

Not Available

#### Recommended material(s)

#### GLOVE SELECTION INDEX

Glove selection is based on a modified presentation of the:

## "Forsberg Clothing Performance Index".

The effect(s) of the following substance(s) are taken into account in the *computer-generated* selection:

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Material	СРІ
PE/EVAL/PE	A
PVA	A
TEFLON	A
BUTYL	С

<sup>\*</sup> CPI - Chemwatch Performance Index

- A: Best Selection
- B: Satisfactory; may degrade after 4 hours continuous immersion C: Poor to Dangerous Choice for other than short term immersion

NOTE: As a series of factors will influence the actual performance of the glove, a final selection must be based on detailed observation. -

\* Where the glove is to be used on a short term, casual or infrequent basis, factors such as "feel" or convenience (e.g. disposability), may dictate a choice of gloves which might otherwise be unsuitable following long-term or frequent use. A qualified practitioner should be consulted.

## **SECTION 9 PHYSICAL AND CHEMICAL PROPERTIES**

#### Information on basic physical and chemical properties

Annooronge	Clear highly flammable liquid with a abarracteristic adoug does not mix		
Appearance	Clear highly flammable liquid with a characteristic odour; does not mix		
Physical state	Liquid	Relative density (Water = 1)	0.94
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	435
pH (as supplied)	Not Available	Decomposition temperature	Not Available
Melting point / freezing point (°C)	-48	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Applicable
Flash point (°C)	13	Taste	Not Available
Evaporation rate	3.1 BuAC = 1	Explosive properties	Not Available
Flammability	HIGHLY FLAMMABLE.	Oxidising properties	Not Available
Upper Explosive Limit (%)	12.5	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	2.1	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	3.9	Gas group	Not Available
Solubility in water (g/L)	1.5%	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	3.45	VOC g/L	668.99

## **SECTION 10 STABILITY AND REACTIVITY**

Reactivity	See section 7
Chemical stability	<ul> <li>Material contains a stabiliser / polymerisation inhibitor system that provides workable but not indefinite shelf life.</li> <li>Storage at higher temperatures and long term storage may result in polymerisation with solidification. In larger quantities e.g. 200 l drums, this may result in generation of heat (exotherm) which may release highly irritating hot vapour.</li> <li>DO NOT open hot exotherming drums - cool externally with water to avoid vapour release.</li> <li>Polymerisation may occur at elevated temperatures and in the presence of ignition sources.</li> <li>Polymerisation may occur slowly at room temperature.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

#### **SECTION 11 TOXICOLOGICAL INFORMATION**

# Information on toxicological effects

Inhaled	Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual.  The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage.  If exposure to highly concentrated vapour atmosphere is prolonged this may lead to narcosis, unconsciousness, even coma and unless resuscitated - death.  Acute effects from inhalation of high vapour concentrations may be chest and nasal irritation with coughing, sneezing, headache and even nausea.
Ingestion	Considered an unlikely route of entry in commercial/industrial environments. If swallowed, the liquid can produce stomach upset, nausea, pain and vomiting.
Skin Contact	Skin contact with the material may damage the health of the individual; systemic effects may result following absorption.  This material can cause inflammation of the skin on contact in some persons.  The material may accentuate any pre-existing dermatitis condition

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Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. Sensitisation reactions may appear suddenly after repeated symptom free exposures This material can cause eye irritation and damage in some persons Eye The vapour is discomforting Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis Chronic There has been some concern that this material can cause cancer or mutations but there is not enough data to make an assessment. There is some evidence that inhaling this product is more likely to cause a sensitisation reaction in some persons compared to the general population. Skin contact with the material is more likely to cause a sensitisation reaction in some persons compared to the general population. Sensitisation may give severe responses to very low levels of exposure, i.e. hypersensitivity. Sensitisation may result in allergic dermatitis responses including rash, itching, hives or swelling of extremities. TOXICITY IRRITATION Lucitone Liquid Not Available Not Available TOXICITY IRRITATION Dermal (rabbit) LD50: >5000 mg/kg<sup>[2]</sup> Eye (rabbit): 150 mg methyl methacrylate Inhalation (rat) LC50: 78 mg/L/4hr<sup>[2]</sup> Skin (rabbit): 10000 mg/kg (open) Oral (rat) LD50: 7872 mg/kg<sup>[2]</sup> IRRITATION ethylene glycol dimethacrylate Oral (rat) LD50: 3300 mg/kg<sup>[2]</sup> Not Available Legend: 1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2.\* Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances MMA is absorbed after inhalation, oral intake and less readily through the skin. Following inhalation it is partly deposited in the airway where it is metabolised by local enzymes. Acute toxicity is low. Skin, eye and airway irritation can result as well as degeneration of the smell function of the nose. The substance is classified by IARC as Group 3: **METHYL METHACRYLATE** NOT classifiable as to its carcinogenicity to humans. Evidence of carcinogenicity may be inadequate or limited in animal testing. Inhalation (human) TCLo: 60 mg/m3(15 ppm) [\* Manuf. Rohm & Haas] UV (ultraviolet)/ EB (electron beam) acrylates are generally of low toxicity UV/EB acrylates are divided into two groups; "stenomeric" and "eurymeric" acrylates. The first group consists of well-defined acrylates which can be described by a simple idealised chemical; they are low molecular weight species with a very ETHYLENE GLYCOL narrow weight distribution profile. DIMETHACRYLATE The eurymeric acrylates cannot be described by an idealised structure and may differ fundamentally between various suppliers; they are of relatively high molecular weigh and possess a wide weight distribution. Stenomeric acrylates are usually more hazardous than the eurymeric substances The following information refers to contact allergens as a group and may not be specific to this product. METHYL METHACRYLATE Contact allergies quickly manifest themselves as contact eczema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact eczema involves & ETHYLENE GLYCOL a cell-mediated (T lymphocytes) immune reaction of the delayed type. Other allergic skin reactions, e.g. contact urticaria, involve antibody-mediated immune DIMETHACRYLATE reactions Asthma-like symptoms may continue for months or even years after exposure to the material ceases. This may be due to a non-allergenic condition known as METHYL METHACRYLATE reactive airways dysfunction syndrome (RADS) which can occur following exposure to high levels of highly irritating compound. Key criteria for the diagnosis & ETHYLENE GLYCOL of RADS include the absence of preceding respiratory disease, in a non-atopic individual, with abrupt onset of persistent asthma-like symptoms within minutes DIMETHACRYLATE to hours of a documented exposure to the irritant. A reversible airflow pattern, on spirometry, with the presence of moderate to severe bronchial hyperreactivity on methacholine challenge testing and the lack of minimal lymphocytic inflammation, without eosinophilia, have also been included in the criteria for diagnosis of RADS Where no "official" classification for acrylates and methacrylates exists, there has been cautious attempts to create classifications in the absence of contrary METHYL METHACRYLATE evidence. For example & ETHYLENE GLYCOL Monalkyl or monoarylesters of acrylic acids should be classified as R36/37/38 and R51/53 DIMETHACRYLATE Monoalkyl or monoaryl esters of methacrylic acid should be classified as R36/37/38 Based on the available oncogenicity data and without a better understanding of the carcinogenic mechanism the Health and Environmental Review Division METHYL METHACRYLATE (HERD), Office of Toxic Substances (OTS), of the US EPA previously concluded that all chemicals that contain the acrylate or methacrylate moiety & ETHYLENE GLYCOL (CH2=CHCOO or CH2=C(CH3)COO) should be considered to be a carcinogenic hazard unless shown otherwise by adequate testing. DIMETHACRYLATE This position has now been revised and acrylates and methacrylates are no longer de facto carcinogens. **Acute Toxicity** Carcinogenicity 0 Skin Irritation/Corrosion Reproductivity 0 Serious Eye STOT - Single Exposure Damage/Irritation Respiratory or Skin 0 STOT - Repeated Exposure sensitisation 0 0 Mutagenicity **Aspiration Hazard** 

X - Data available but does not fill the criteria for classification

Legend:

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— Data required to make classification available

O - Data Not Available to make classification

#### **SECTION 12 ECOLOGICAL INFORMATION**

#### Toxicity

Ingredient	Endpoint	Test Duration (hr)	Species	Value	Source
methyl methacrylate	LC50	96	Fish	43.382mg/L	3
methyl methacrylate	EC50	48	Crustacea	=69mg/L	1
methyl methacrylate	EC50	72	Algae or other aquatic plants	>110mg/L	2
methyl methacrylate	EC3	192	Algae or other aquatic plants	=37mg/L	1
methyl methacrylate	NOEC	504	Crustacea	37mg/L	2
ethylene glycol dimethacrylate	LC50	96	Fish	18.578mg/L	3
ethylene glycol dimethacrylate	EC50	96	Algae or other aquatic plants	10.1mg/L	2
ethylene glycol dimethacrylate	EC50	504	Crustacea	>5.05mg/L	2
ethylene glycol dimethacrylate	NOEC	96	Algae or other aquatic plants	0.804mg/L	2
Legend:	Aquatic Toxicity Da	Extracted from 1. IUCLID Toxicity Data 2. Europe ECHA Registered Substances - Ecotoxicological Information - Aquatic Toxicity 3. EPIWIN Suite V3.12 - Aquatic Toxicity Data (Estimated) 4. US EPA, Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data			

DO NOT discharge into sewer or waterways.

#### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
methyl methacrylate	LOW	LOW
ethylene glycol dimethacrylate	LOW	LOW

#### **Bioaccumulative potential**

Ingredient	Bioaccumulation
methyl methacrylate	LOW (BCF = 6.6)
ethylene glycol dimethacrylate	LOW (LogKOW = 2.2088)

## Mobility in soil

Ingredient	Mobility
methyl methacrylate	LOW (KOC = 10.14)
ethylene glycol dimethacrylate	LOW (KOC = 27.15)

# **SECTION 13 DISPOSAL CONSIDERATIONS**

# Waste treatment methods

Product / Packaging

- ▶ Consult manufacturer for recycling options and recycle where possible .
- ► Consult State Land Waste Management Authority for disposal.
- ► Incinerate residue at an approved site.
- ▶ Recycle containers if possible, or dispose of in an authorised landfill.
- ► Containers may still present a chemical hazard/ danger when empty.
- ▶ Return to supplier for reuse/ recycling if possible.

#### Otherwise:

- If container can not be cleaned sufficiently well to ensure that residuals do not remain or if the container cannot be used to store the same product, then puncture containers, to prevent re-use, and bury at an authorised landfill.
- ▶ Where possible retain label warnings and SDS and observe all notices pertaining to the product.

## **SECTION 14 TRANSPORT INFORMATION**

disposal

#### **Labels Required**



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#### **Lucitone Liquid**

UN number	1247		
UN proper shipping name	METHYL METHACRYLATE MONOMER, STABILISED		
Transport hazard class(es)	Class 3 Subrisk Not Applicable		
Packing group	П		
Environmental hazard	Not Applicable		
Special precautions for user	Special provisions Not Applicable  Limited quantity 1 L		

• ` `			
UN number	1247		
UN proper shipping name	Methyl methacrylate monomer, stabilized		
Transport hazard class(es)	ICAO/IATA Class 3 ICAO / IATA Subrisk Not Applicable ERG Code 3L		
Packing group	П		
Environmental hazard	Not Applicable		
	Special provisions  Cargo Only Packing Instructions	Not Applicable 364	
	Cargo Only Maximum Qty / Pack	60 L	
Special precautions for user	Passenger and Cargo Packing Instructions	353	
	Passenger and Cargo Maximum Qty / Pack	5L	
	Passenger and Cargo Limited Quantity Packing Instructions	Y341	
	Passenger and Cargo Limited Maximum Qty / Pack	1L	

#### Sea transport (IMDG-Code / GGVSee)

UN number	1247	
UN proper shipping name	METHYL METHACRYLATE MONOMER, STABILIZED	
Transport hazard class(es)	IMDG Class 3 IMDG Subrisk Not Applicable	
Packing group		
Environmental hazard	Not Applicable	
Special precautions for user	EMS Number F-E, S-D Special provisions Not Applicable Limited Quantities 1 L	

#### Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

## **SECTION 15 REGULATORY INFORMATION**

## Safety, health and environmental regulations / legislation specific for the substance or mixture

#### METHYL METHACRYLATE(80-62-6) IS FOUND ON THE FOLLOWING REGULATORY LISTS

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Australia Exposure Standards Australia Hazardous Substances Information System - Consolidated Lists Monographs International Air Transport Association (IATA) Dangerous Goods Regulations - Prohibited List Australia Inventory of Chemical Substances (AICS) Passenger and Cargo Aircraft

#### ETHYLENE GLYCOL DIMETHACRYLATE(97-90-5) IS FOUND ON THE FOLLOWING REGULATORY LISTS

Australia Hazardous Substances Information System - Consolidated Lists		Australia Inventory of Chemical Substances (AICS)
National Inventory	Status	
Australia - AICS	Y	
Canada - DSL	Υ	
Canada - NDSL	N (methyl methacrylate; ethylene glycol dimethacrylate)	
China - IECSC	Υ	
Europe - EINEC / ELINCS / NLP	Y	

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Japan - ENCS	Y
Korea - KECI	Υ
New Zealand - NZIoC	Υ
Philippines - PICCS	Υ
USA - TSCA	Υ
Legend:	Y = All ingredients are on the inventory N = Not determined or one or more ingredients are not on the inventory and are not exempt from listing(see specific ingredients in brackets)

#### **SECTION 16 OTHER INFORMATION**

#### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references

A list of reference resources used to assist the committee may be found at:

www.chemwatch.net

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios. Scale of use, frequency of use and current or available engineering controls must be considered.

#### **Definitions and abbreviations**

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit,

IDLH: Immediately Dangerous to Life or Health Concentrations

OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value

LOD: Limit Of Detection

OTV: Odour Threshold Value

BCF: BioConcentration Factors

BEI: Biological Exposure Index

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