# **Hydrochloric Acid, 7.7N**

## Safety Data Sheet



# Section 1 Product Description

Product Name: Hydrochloric Acid, 7.7N

Manufacturer number HA6207-B
Distributor: Best Sanitizers, Inc.

PO Box 1360 Penn Valley, CA 95946

**Chemical Information Emergency:** 

Aquaphoenix Scientific 1.800.255.3924

# Section 2 Hazard Identification

Classification of the substance or mixture:

Specific target organ toxicity following single exposure

Acute Toxicity (oral,dermal,inhalation)

Serious eye damage

Category 1

Skin corrosion

Category 1A

Corrosive to metals

Category 1

### **Danger**







Appearance—Aqueous solution
Physical state—Liquid
Odor— Pungent Odor

#### **Hazard Statements**

May be corrosive to metals. Causes severe skin burns and eye damage.

Causes serious eye damage.

Toxic if swallowed.

May cause drowsiness or dizziness.

#### **Precautionary Statements**

If medical advice is needed, have product container or label at hand.

Keep out of reach of children.

Read label before use. Do not breathe dust/fume/gas/mist/vapours/spray.

Wash skin thoroughly after handling.

Wear protective gloves/protective clothing/eye protection/face protection.

Do not eat, drink or smoke when using this product.

Avoid breathing dust/fume/gas/mist/vapours/spray.

Use only outdoors or in a well-ventilated area.

Keep only in original container.

IF ON SKIN (or hair): Remove/Takeoff immediately all contaminated clothing. Rinse skin with water/shower. Wash contaminated clothing.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

IF SWALLOWED: Rinse mouth. DO NOT induce vomiting. Call a POISON CENTER or doctor if you feel unwell.

Specific treatment (see Section 4).

In case of fire: Use media appropriate for extinction. Store locked up.

Store in a well ventilated place. Store in corrosive resistant stainless steel container with a resistant inner liner. Dispose of contents/container according to local, regional, national, territorial, provincial, and international regulation.

### **Precautionary Statements—Disposal**

Dispose of contents/container to an approved waste disposal plant.

#### Hazards not otherwise classified (HNOC)

None.

#### Other Information

No ingredients of Unknown Acute Toxicity.

Section 3	Composition/Information on Ingredients
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Chemical Name	CAS No.	Weight-%
Hydrochloric Acid, ACS	7647-01-0	75.05
Water	7732-18-5	24.95

Section 4	First Aid Measures
First Aid Measures Eye Contact	Hold eye(s) open and rinse slowly and gently with water for 15-20 minutes. Remove contact lenses, if present, after first 5 minutes, then continue rinsing eye(s). Get medical advice/attention.
Skin Contact	Wash affected area with soap and water. Rinse/flush exposed skin gently using water for 15-20 minutes. Get medical attention if irritation persists or if concerned.
Inhalation	Remove to fresh air. Seek immediate medical attention if discomfort or irritation persists.
Ingestion	Rinse mouth thoroughly. Do NOT induce vomiting. Drink sips of water. Seek medical attention if irritation, discomfort or vomiting persists.

# Most important symptoms and effects, both acute and delayed Symptoms

Inhalation may cause irritation to nose and upper respiratory tract, ulceration, coughing, chest tightness and shortness of breath. Higher concentrations cause tachypnoea, pulmonary oedema and suffocation. Ingestion may cause corrosion of lips, mouth, oesophagus and stomach, dysphagia and vomiting. Pain, eye, ulceration, conjunctival irritation, cataracts and glaucoma may occur following eye exposure. Erythema and skin irritation, as well as chemical burns to skin and mucous membranes may arise following skin eposure. Potential sequelae following ingestion of hydrochloric acid include perforation, scarring of the esophagus or stomach obstruction. In some cases, RADS may develop. Respiratory symptoms may take up to 36 hours to develop.

#### Indication of any immediate medical attention and special treatment needed

If seeking medical attention, provide SDS document to physican.

### **Section 5**

### **Fire-Fighting Measures**

#### **Suitable Extinguishing Media**

If in laboratory setting, follow laboratory fire suppression procdures. Use appropriate suppression agents for adjacent combustible materials or sources of ignition.

#### **Unsuitable Extinguishing Media**

None.

#### Specific hazards arising from the chemical

Combustion products may include carbon oxides or other toxic vapors. Hydrogen chloride gas.

#### **Protective Equipment and Precautions for Firefighters**

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

Thermal decomposition can produce poisoning chlorine. Hydrochloric acid reacts also with many organic materials with liberation of heat.

### **Section 6**

### **Accidental Release Measures**

### Personal Precautions, Protective Equipment and Emergency Procedures

Personal Precautions Use personal protection recommended in Section 8. Ensure adequate ventilation, especially in

confined areas. Keep unprotected persons away. Keep away from ignition sources. Protect

from heat. Contained spilled material by diking or using inert absorbent.

**Environment Precautions** 

Environmental Precautions Prevent entry into waterways, sewers, basements or confined areas. See Section 12 for additional

ecological information.

Methods and material for containment and cleaning up

Methods for containment Prevent further leakage or spillage if safe to do so. Contain and collect spillage with non-combustible

absorbent material, (e.g. sand, earth, diatomaceous earth, vermiculite) and place in container

for disposal according to local/national regulations (See Section 13).

Methods for cleaning up Use clean non-sparking tools to collect absorbed material. May be ignited by friction, heat, sparks

or flames. Collect spillage. Soak up with inert absorbent material. Sweep up and shovel into suitable containers for disposal. Following product recovery, flush area with water. Place in properly labeled containers. If in a laboratory setting, follow Chemical Hygiene Plan procedures.

## **Section 7**

# **Handling and Storage**

#### **Precautions for Safe Handling**

**Advice on Safe Handling** 

Prevent formation of aerosols. If opening metal containers, use non-sparking tools because of the possibility of hydrogen gas being preent. Wash hands after handling. Avoid contact with skin, eyes and clothing. Do not eat, drink or smoke when using this product. Use personal protection recommended in Section 8. When handling hydrochloric acid avoid contact with metals and organic matters. Never use hot water and never add water to the acid!

#### Conditions for safe storage, including any incompatibilities

Storage Conditions/ Incompatible materials Keep Containers tightly closed in a dry, cool and well-ventilated place. Avoid storage near extreme heat, ignition sources or open flame. Keep away from foodstuffs. Store from oxidizing agents. Protect from freeaiwg and physical damage. Containers for hydrochloric acid must be made from corrosion resistant materials: glass, polyethylene, polypropylene, polyvinyl chloride, carbon steel lined with rubber or ebonite.

### Section 8 Protection Information

**Exposure Guidelines** 

Respiratory protection

**General Hygiene** 

Control Parameters: 7647-01-0 Hydrochloric Acid, ACGIH: 2 ppm Ceiling

7647-01-0, Hydrochloric Acid, NIOSH: 5 ppm Ceiling; 7 mg/m<sup>3</sup> Ceiling.

7647-01-0 Hydrochloric Acid, OSHA PEL TWA 7 mg/m<sup>3</sup>

**Appropriate Engineering Controls** 

Engineering Controls Emergency eyewash fountains and safety showers should be available in the immediate vicinity of use/

handling. Provide exhaust ventilation or other engineering controls to keep the airborne concentrations

of vapor or mists below the applicable workplace exposure limits.

#### Individual Protection Measures, such as personal protective equipment

Eye/Face protection Wear safety glasses with side shields (or goggles).

Skin and body protection Wear impermeable and resistant to the product/ substance/preparation protective gloves. Selection

of glove material on consideration of the penetration times, rates of diffusion and degradation.

If exposure limits are exceeded or irritation is experienced, NIOSH/MSHA approved respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high

airborne contaminant concentrations. Respiratory protection must be provided in accordance with current local regulations.

The usual precautionary measures are to be adhered to when handling chemicals. Keep away from food, beverages and feed sources. Immediately remove all soiled and contaminated clothing. Wash face, hands and any exposed skin thoroughly after handling. Wash contaminated

clothing and shoes before reuse. Do not Eat, Drink or Smoke when using this product. Do not inhale

gases/fumes/dust/mist/vapors/aerosols. Avoid contact with the eyes and skin.

## Section 9

# **Physical and Chemical Properties**

#### Information on basic physical and chemical properties

Formula: See Section 3 Physical State: Liquid

Odor: Pungent odor Appearance: Aqueous solution

Odor Threshold: 0.3 - 14.9 mg/m<sup>3</sup> Color: Clear

Property Values
pH < 1

Meting Point/Freezing Point

Approximately 0°C

Boiling Point/Boiling Range

Approximately 100°C

Flash Point Not Applicable.
Evaporation rate > 1.00

Flammability (solid, gas) Non Combustible

Flammability Limit in Air

Upper flammability limit:
Lower flammability limit:
Non Explosive
Not Determined
Vapor density:
Not Determined
Specific Gravity
Not Determined
Water solubility
Soluble in water

Partition coefficient No information available

Autoignition temperature Not Determined

Decomposition temperature Not Determined

Kinematic viscosity

No information available

Dynamic viscosity

No information available

# Section 10 Stability and Reactivity Data

**Reactivity** Reacts violently with bases and is corrosive.

Chemical Stability No decomposition if used and stored according to specifications.

Possibility of Hazardous Reactions Reacts violently with oxidants forming toxic gas (chlorine). Attacks many metals in the

presence of water forming flammable/explosive gas (hydrogen).

Conditions to avoid Excess heat. Incompatible products.

Incompatible materials Metal oxides, formaldehyde, Strong bases, Most metals, Stong oxidizing agents, Reducing agents.

Alkalis, cyanides, sulfides. Sulfites.

**Hazardous Decomposition Products**Carbon oxides (CO, CO<sub>2</sub>). Fumes of hydrogen chloride and hydrogen in contact with metals.

Oxides of carbon.

## Section 11 Toxicity Data

#### Information on toxicological effects

Inhalation Hydrochloric acid

Sensitization No Information Available
Germ cell mutagenicity No Information Available

Carcinogenicity IARC: Group 3

Reproductive toxicity
STOT single exposure
STOT repeated exposure
No Information Available
No Information Available
No Information Available

# **Section 12**

# **Ecological Data**

### **Ecotoxicity**

None.

### Persistence and degradability

Readily Biodegradable.

#### **Bioaccumulation**

Not Bioaccumulative.

### **Mobility**

Aqueous solution has high mobility in soil.

### Other adverse effects

None.

## Section 13

# **Disposal Information**

#### Waste treatment methods

Disposal of wastes

Cover spill with soda ash or calcium carbonate. Mix and add water to form slurry. Decant to drain. Treat the solid residue as normal refuse. All chemical waste generators must determine whether a discarded chemical is classified as hazardous waste. Disposal should be in accordance with applicable regional, national and local laws and regulations. Do not dispose together with household garbage.

### Section 14

# Transport Information

#### DOT

UN ID Number UN1789

UN proper shipping name HYDROCHLORIC ACID

Hazard Class 8 Packing Group II

### **Section 15**

### **Regulatory Information**

**International Inventories** 

TSCA Complies
DSL/NDSL Complies

#### Legend:

TSCA—United States Toxic Substances Control Act Section 8(b) Inventory

DSL/NDSL— Canadian Domestic Substances List/Non-Domestic Substances List

#### **US Federal Regulations**

**SARA 313** 

Section 313 of Title III of the Superfund Amendments and Reauthorization of 1986 (SARA).

7647-01-0 Hydrochloric Acid

#### **SARA 311/312**

Acute

#### **CERCLA**

7647-01-0 Hydrochloric Acid 5000

#### **US State Regulations**

#### **California Proposition 65**

None of the ingredients is listed.

Section 2	16	Addit		
<u>NFPA</u>	Health Hazards 3	Flammability 0	Instability 1	Physical and Chemical Properties  O
<u>HMIS</u>	Health Hazards	Flammability	Physical Hazards	Personal protection
	3	0	1	X
Prepared by:	Technical Department			

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#### Disclaimer

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