

### **MATERIAL SAFETY DATA SHEET**

## PRODUCT NAME: SULFUR HEXAFLUORIDE

## 1. Product and Company Identification

BOC Gases,
Division of
BOC Gases
Division of,

The BOC Group, Inc.

575 Mountain Avenue

Murray Hill, NJ 07974

BOC Canada Limited

5975 Falbourne Street, Unit 2

Mississauga, Ontario L5R 3W6

**TELEPHONE NUMBER:** (908) 464-8100 **TELEPHONE NUMBER:** (905) 501-1700

24-HOUR EMERGENCY TELEPHONE NUMBER: 24-HOUR EMERGENCY TELEPHONE NUMBER:

CHEMTREC (800) 424-9300 (905) 501-0802

**EMERGENCY RESPONSE PLAN NO: 2-0101** 

PRODUCT NAME: SULFUR HEXAFLUORIDE

CHEMICAL NAME: Sulfur hexafluoride COMMON NAMES/SYNONYMS: None TDG (Canada) CLASSIFICATION: 2.2

WHMIS CLASSIFICATION: A

PREPARED BY: Loss Control (908)464-8100/(905)501-1700

PREPARATION DATE: 6/1/95 REVIEW DATES: 07/16/04

# 2. Composition, Information on Ingredients

#### **EXPOSURE LIMITS<sup>1</sup>:**

INGREDIENT	% VOLUME	PEL-OSHA <sup>2</sup>	TLV-ACGIH <sup>3</sup>	LD <sub>50</sub> or LC <sub>50</sub> Route/Species
Sulfur hexafluoride FORMULA: SF <sub>6</sub> CAS: 2551-62-4 RTECS #: WS4900000	99.9	1000 ppm TWA	1000 ppm TWA	Not Available

<sup>&</sup>lt;sup>1</sup> Refer to individual state or provincial regulations, as applicable, for limits which may be more stringent than those listed here.

OSHA Regulatory Status: This material is classified as hazardous under OSHA regulations.

## 3. Hazards Identification

#### **EMERGENCY OVERVIEW**

Odorless, colorless, non-flammable gas. Simple Asphyxiant - This product does not contain oxygen and may cause asphyxia if released in a confined area. Maintain oxygen levels above 19.5%. Decomposes to toxic fluoride compounds at temperatures above 400  $^{\circ}$ F (204  $^{\circ}$ C). Contents under pressure. Use and store below 125  $^{\circ}$ F.

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<sup>&</sup>lt;sup>2</sup> As stated in 29 CFR 1910, Subpart Z (revised July 1, 1993)

<sup>&</sup>lt;sup>3</sup> As stated in the ACGIH 2004 Threshold Limit Values for Chemical Substances and Physical Agents.

### **ROUTE OF ENTRY:**

Skin Contact	Skin Absorption	Eye Contact	Inhalation	Ingestion
No	No	No	Yes	No

#### **HEALTH EFFECTS:**

Exposure Limits	Irritant	Sensitization
Yes	No	No
Teratogen	Reproductive Hazard	Mutagen
No	No	No
Synergistic Effects		
None reported		

Carcinogenicity: -- NTP: No IARC: No OSHA: No

**EYE EFFECTS:** None known for gas. Contact with rapidly expanding gas near the point of release may cause frostbite.

**SKIN EFFECTS:** None known for gas. Contact with rapidly expanding gas near the point of release may cause frostbite with redness, skin color change to gray or white, and blistering.

**INGESTION EFFECTS:** None known. Ingestion is unlikely as product is gas at room temperature.

**INHALATION EFFECTS:** Effects of oxygen deficiency resulting from simple asphyxiants may include: rapid breathing, diminished mental alertness, impaired muscular coordination, faulty judgement, depression of all sensations, emotional instability, and fatigue. As asphyxiation progresses, nausea, vomiting, prostration, and loss of consciousness may result, eventually leading to convulsions, coma, and death.

MEDICAL CONDITIONS AGGRAVATED BY EXPOSURE: None known.

POTENTIAL ENVIRONMENTAL EFFECTS: Not expected to exert an ecotoxic effect.

### 4. First Aid Measures

**EYES:** None required for gas. If frostbite is suspected, flush eyes with cool water for 15 minutes and obtain immediate medical attention.

**SKIN:** None required for gas. For frostbite, immerse skin in lukewarm water. DO NOT USE HOT WATER. Obtain medical attention.

**INGESTION:** None required.

**INHALATION:** PROMPT MEDICAL ATTENTION IS MANDATORY IN ALL CASES OF OVEREXPOSURE. RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS. Victims should be assisted to an uncontaminated area and inhale fresh air. Quick removal from the contaminated area is most important. Unconscious persons should be moved to an uncontaminated area, and if breathing has stopped, administer artificial resuscitation and supplemental oxygen. Further treatment should be symptomatic and supportive.

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## 5. Fire Fighting Measures

Conditions of Flammability: Nonflammable				
Flash point:	Method:		Autoignition	
None	Not Applicable		Temperature: None	
LEL(%): None		UEL(%): None		
Hazardous combustion products: Oxides of sulfur, fluorine compounds				
Sensitivity to mechanical shock: None				
Sensitivity to static discharge: None				

**FIRE AND EXPLOSION HAZARDS:** Nonflammable. Decomposes to highly toxic sulfur oxides and fluorine compounds under fire conditions. Cylinders may vent rapidly or rupture violently from pressure when involved in a fire situation.

**EXTINGUISHING MEDIA:** None required. Use as appropriate for surrounding materials.

#### FIRE FIGHTING INSTRUCTIONS:

Fire fighters should wear a full facepiece self-contained breathing apparatus (SCBA) operated in positive pressure mode and full turnout or bunker gear. Continue to cool fire exposed cylinders until well after flames are extinguished.

### 6. Accidental Release Measures

Evacuate all personnel from affected area. Use appropriate protective equipment (See Section 8). Stop the flow of gas or remove cylinder to outdoor location if this can be done without risk. Ventilate enclosed areas. If leak is in user's equipment, be certain to purge piping with inert gas prior to attempting repairs. If leak is in container or container valve, contact the appropriate emergency telephone number listed in Section 1 or call your closest BOC location.

## 7. Handling and Storage

**Electrical classification:** Non-hazardous.

This gas mixture is noncorrosive and may be used with all common structural materials.

Aluminum, stainless steel, copper, brasses or silver recommended for temperatures above 400 °F (204 °C) to prevent decomposition to toxic fluoride compounds.

Use only in well-ventilated areas. Valve protection caps must remain in place unless container is secured with valve protection outlet piped to use point. Do not drag, slide or roll cylinders. Use a suitable hand truck for cylinder movement. Use a pressure-reducing regulator when connecting cylinder to lower pressure piping or systems. Do not heat cylinder by any means to increase the discharge rate of product from the cylinder. Use a check valve or trap in the discharge line to prevent hazardous back flow into the cylinder. Do not insert any object (i.e.: screwdriver) into valve cap openings as this can damage the valve causing leakage.

Protect cylinders from physical damage. Store in cool, dry, well-ventilated area of non-combustible construction away from heavily trafficked areas and emergency exits. Do not allow the temperature where cylinders are stored to exceed 125 °F (52 °C). Cylinders should be stored upright and firmly secured to prevent falling or being knocked over. Use a "first in-first out" inventory system to prevent full cylinders being stored for excessive periods of time.

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For additional recommendations, consult Compressed Gas Association Pamphlets P-1, P-14, and Safety Bulletin SB-2.

Never carry a compressed gas cylinder or a container of a gas in cryogenic liquid form in an enclosed space such as a car trunk, van or station wagon. A leak can result in a fire, explosion, asphyxiation or a toxic exposure.

### 8. Exposure Controls, Personal Protection

**ENGINEERING CONTROLS:** Use local exhaust in combination with general ventilation as necessary to control air contaminants at or below acceptable exposure guidelines and maintain oxygen levels in the air at or above 19.5%.

**EYE/FACE PROTECTION:** Safety goggles or glasses.

**SKIN PROTECTION:** Protective gloves appropriate for the job.

**RESPIRATORY PROTECTION:** For emergency release and conditions with exposures above the applicable exposure limits use a positive pressure NIOSH approved air-supplying respirator system (SCBA or airline/escape bottle) using a minimum Grade D air.

OTHER/GENERAL PROTECTION: Safety shoes, emergency eyewash station

## 9. Physical and Chemical Properties

PARAMETER	VALUE	UNITS	
Physical state (gas, liquid, solid)	: Gas		
Vapor pressure	: 319.1	psia	
Vapor density (Air = 1)	: 5.13		
Evaporation Point	: Not Available		
Boiling point	: -82.8	°F	
	: -63.8	°C	
Freezing point	: -59.4	°F	
	: -50.8	°C	
PH	: Not Applicable		
Specific gravity	: Not Available		
Oil/water partition coefficient	: Not Available		
Solubility (H <sub>2</sub> 0)	: Slight		
Odor threshold	: Not Applicable		
Odor and appearance	: Colorless, odorless gas	S	

# 10. Stability and Reactivity

**STABILITY:** Stable below 400 °F (204 °C).

**INCOMPATIBLE MATERIALS/CONDITIONS:** Reported to explode in contact with disilane. Oxygen and certain metals cause slow decomposition to toxic fluorides.

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**HAZARDOUS DECOMPOSITON PRODUCTS:** Thionyl fluoride compounds including  $SF_2$ ,  $SF_4$ ,  $S_2F_2$  and  $S_2F_{10}$ .

HAZARDOUS POLYMERIZATION: Does not occur.

## 11. Toxicological Information

**INHALATION:** Sulfur hexafluoride is considered to be chemically and physiologically inert in its pure state. Fifty rats exposed to an atmosphere consisting of 80% sulfur hexafluoride and 20% oxygen for 16 to 25 hours exhibited no effects from exposure.

**SKIN AND EYE:** Does not cause skin and eye irritation.

**OTHER:** Due to the low toxicity of sulfur hexafluoride, chronic studies have not been conducted.

Oxygen deficiency during pregnancy has produced developmental abnormalities in humans and experimental animals.

## 12. Ecological Information

Does not contain Class I or Class II ozone depleting substances. Non-toxic. Not expected to bioconcentrate.

## 13. Disposal Considerations

Do not attempt to dispose of residual waste or unused quantities. Return in the shipping container PROPERLY LABELED, WITH ANY VALVE OUTLET PLUGS OR CAPS SECURED AND VALVE PROTECTION CAP IN PLACE to BOC Gases or authorized distributor for proper disposal.

### 14. Transport Information

PARAMETER	United States DOT	Canada TDG
PROPER SHIPPING NAME:	Sulfur Hexafluoride	Sulfur Hexafluoride
HAZARD CLASS:	2.2	2.2
IDENTIFICATION NUMBER:	UN 1080	UN 1080
SHIPPING LABEL:	NONFLAMMABLE GAS	NONFLAMMABLE GAS

# 15. Regulatory Information

SARA TITLE III NOTIFICATIONS AND INFORMATION SARA TITLE III - HAZARD CLASSES:

Sudden Release of Pressure Hazard

#### SARA TITLE III - SECTION 313 SUPPLIER NOTIFICATION:

This product does not contain toxic chemicals subject to the reporting requirements of section 313 of the Emergency Planning and Community Right-To-Know Act (EPCRA) of 1986 and of 40 CFR 372.

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U.S. TSCA/Canadian DSL: All ingredients are listed on the U.S. Toxic Substances Control Act (TSCA) inventory or exempt from listing and on the Canadian Domestic Substance List (DSL).

California Proposition 65: This product does not contain ingredient(s) known to the State of California to cause cancer or reproductive toxicity.

Canadian Controlled Products Regulations (CPR): This product has been classified in accordance with the hazard criteria of the CPR and the MSDS contains all the information required by the CPR.

### 16. Other Information

NFPA HAZA	ARD CODES	HMIS HAZARI	O CODES	RATINGS SYSTEM
Health: Flammability: Instability:		Health: Flammability: Physical Hazard:		0 = No Hazard 1 = Slight Hazard 2 = Moderate Hazard 3 = Serious Hazard 4 = Severe Hazard

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Note: Ratings were assigned in accordance with Compressed Gas Association (CGA) guidelines as published in CGA Pamphlet P-19-2004, CGA Recommended Hazard Ratings for Compressed Gases, 2<sup>nd</sup> Edition.

American Conference of Governmental Industrial Hygienists **ACGIH** DOT Department of Transportation **IARC** International Agency for Research on Cancer NTP National Toxicology Program **OSHA** Occupational Safety and Health Administration PEL Permissible Exposure Limit **SARA** Superfund Amendments and Reauthorization Act STEL Short Term Exposure Limit

TDG Transportation of Dangerous Goods

TLV Threshold Limit Value

WHMIS Workplace Hazardous Materials Information System

Compressed gas cylinders shall not be refilled without the express written permission of the owner. Shipment of a compressed gas cylinder which has not been filled by the owner or with his/her (written) consent is a violation of transportation regulations.

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