

SAFETY DATA SHEET

044

Product Name R409B

1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

Supplier Name BOC LIMITED (AUSTRALIA)

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Synonym(s) 044 - SDS NUMBER • CHLORODIFLUOROETHANE (HCFC142B) BLEND • CHLORODIFLUOROMETHANE

(HCFC22) • CHLOROTETRAFLUOROETHANE (HCFC124) • FORANE FX57 • R409 B

Use(s) REFRIGERANT SDS Date 26 Mar 2010

2. HAZARDS IDENTIFICATION

NOT CLASSIFIED AS HAZARDOUS ACCORDING TO ASCC CRITERIA

CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE

UN No. 1078 DG Class 2.2 Subsidiary Risk(s) None Allocated

Packing GroupNone AllocatedHazchem Code2REEPG2C2

3. COMPOSITION/ INFORMATION ON INGREDIENTS

Ingredient	Formula	CAS No.	Content
CHLORODIFLUOROMETHANE (HCFC-22)	C-H-CI-F2	75-45-6	65%
2-CHLORO-1,1,1,2-TETRAFLUOROETHANE (HCFC-124)	C2-H-CI-F4	2837-89-0	25%
1-CHLORO-1,1-DIFLUOROETHANE (HCFC 142B)	C2-H3-CI-F2	75-68-3	10%

4. FIRST AID MEASURES

Eye Cold burns: Immediately flush with tepid water or with sterile saline solution. Hold eyelids apart and irrigate for 15

minutes. Seek medical attention.

Inhalation If inhaled, remove from contaminated area. To protect rescuer, use an Air-line respirator or Self Contained

Breathing Apparatus (SCBA). Apply artificial respiration if not breathing. Give oxygen if available. For advice,

contact a Poisons Information Centre on 13 11 26 (Australia Wide) or a doctor.

Skin Cold burns: Remove contaminated clothing and gently flush affected areas with warm water (30°C) for 15 minutes.

Apply sterile dressing and treat as for a thermal burn. For large burns, immerse in warm water for 15 minutes. DO

NOT apply any form of direct heat. Seek immediate medical attention.

Ingestion Due to product form and application, ingestion is considered unlikely.

Advice to Doctor Use of adrenaline and other catecholamines may be contraindicated due to possible cardiac sensitisation.

Treatment for asphyxia.



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5. FIRE FIGHTING MEASURES

Flammability Non flammable. May evolve toxic gases (chlorides, phosgene, fluorides, carbon oxides) when heated to

decomposition.

Fire and **Explosion** Temperatures in a fire may cause cylinders to rupture. Cool cylinders or containers exposed to fire by applying

water from a protected location. Do not approach cylinders or containers suspected of being hot.

Extinguishing Use water fog to cool containers from protected area.

Hazchem Code 2RF

6. ACCIDENTAL RELEASE MEASURES

Spillage

Always ensure cylinder pressure is below equipment pressure rating and any relief valve setting. Contact manufacturer for further information. Leak checking may be done by pressure drop test or by using soapy water on outlets and outlets. Shut cylinder valve to stop gas leaks from equipment if possible and safe to do so. If cylinder or cylinder valve is leaking then shut the cylinder valve, depressurise the equipment, disconnect cylinder from equipment and move the cylinder to a well ventilated area, preferably outdoors. Never attempt to repair a leaking or damaged cylinder valve.

7. STORAGE AND HANDLING

Storage

Do not store near incompatible materials. Cylinders should be stored below 45°C in a secure area, upright and restrained to prevent cylinders from falling. Cylinders should also be stored in a dry, well ventilated area constructed of non-combustible material with firm level floor (preferably concrete), away from areas of heavy traffic and emergency exits.

Handling

Use of safe work practices are recommended to avoid eye or skin contact and inhalation. Do not drag, drop, slide or roll cylinders. The uncontrolled release of a gas under pressure may cause physical harm. Use a suitable hand truck for cylinder movement.

8. EXPOSURE CONTROLS/ PERSONAL PROTECTION

Exposure Stds

Ingradiant	Reference	TWA		STEL		
Ingredient		ppm	mg/m3	ppm	mg/m3	
2-CHLORO-1,1,1,2- TETRAFLUOROETHANE (HCFC- 124)	ASCC (AUS)	Asphyxiant				
Chlorodifluoromethane	ASCC (AUS)	1000	3540			

1-CHLORO-1,1-DIFLUOROETHANE (HCFC 142B)

ES-TWA: Asphyxiant WES-TWA: Asphyxiant

Biological Limits No biological limit allocated.

Engineering Controls

Avoid inhalation. Use in well ventilated areas. Where an inhalation risk exists, mechanical extraction ventilation is

recommended. Maintain vapour levels below the recommended exposure standard.

PPF

Wear safety boots, leather gloves, coveralls and safety glasses. Where an inhalation risk exists, wear: an Air-line respirator.









9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance CLEAR COLOURLESS LIQUID Solubility (Water) NOT AVAILABLE Odour SLIGHT ETHEREAL ODOUR **Specific Gravity** NOT APPLICABLE

NOT APPLICABLE % Volatiles pН 100 % Vapour Pressure 850 kPa @ 25°C **Flammability** NON FLAMMABLE Flash Point Vapour Density 3.5 (Air = 1)NOT RELEVANT



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Boiling Point -35.2°C NOT RELEVANT **Upper Explosion Limit Melting Point NOT AVAILABLE Lower Explosion Limit** NOT RELEVANT

Evaporation Rate NOT APPLICABLE

Critical Pressure 3675 kPa **Critical Temperature** 184°C

10. STABILITY AND REACTIVITY

Chemical Stability Stable under recommended conditions of storage.

Conditions to Avoid Avoid heat, sparks, open flames and other ignition sources.

Material to Avoid Incompatible with oxidising agents (eg. hypochlorites), alkalis/ alkali earth metals. May also react violently

> with sodium, potassium, barium and other alkali or alkaline earth metals and finely divided metals. Compounding ingredients in natural rubber can be extracted during rapid liquid withdrawal and will swell.

Decomposition May evolve toxic gases (chlorides, phosgene, fluorides, carbon oxides) when heated to decomposition.

Hazardous Reactions Polymerization will not occur.

11. TOXICOLOGICAL INFORMATION

Health Hazard Summary

Asphyxiant. Symptoms of exposure are directly related to displacement of oxygen. As the amount of oxygen inhaled is reduced from 21-14% volume, the pulse rate may accelerate and the rate and volume of breathing may increase. The ability to maintain attention and think clearly is diminished, muscular co-ordination is somewhat disturbed. As oxygen decreases from 14-10% volume, judgement becomes faulty, severe injuries may result in no pain. Muscular effort may lead to rapid fatigue. Further reduction to 6% may result in nausea and vomiting. Ability to move may be lost. Permanent brain damage may result even after resuscitation from exposure to this low level of oxygen. Below 6% breathing is in gasps and convulsions may occur. Inhalation of a mixture containing no oxygen may result in unconsciousness from the first breath and death may follow in minutes.

Eve Irritant vapour. Low temperature evaporating liquid can cause cold burns.

Inhalation Irritant - asphyxiant. Effects are proportional to oxygen displacement. May cause dizziness, headache, confusion

and unconsciousness at concentrations. Prolonged exposure to high concentrations may result in sensitisation to

the effects of adrenalins on the heart.

Skin Irritating vapour. Direct contact with the liquefied material or escaping compressed gas may cause frost-bite injury.

Ingestion Ingestion is considered unlikely due to product form. However, ingestion may result in discomfort of the

gastrointestinal tract from rapid evaporation of liquid and consequent evolution of gas. Some of the effects of

inhalation would be expected.

CHLORODIFLUOROMETHANE (HCFC-22) (75-45-6) **Toxicity Data**

Carcinogenicity: Not classifiable as to its carcinogenicity (IARC Group 3)

LC50 (Inhalation): 35 pph/15 minutes (rat) LCLo (Inhalation): 25 pph/4 hours (rat)

TCLo (Inhalation): 50000 ppm/5 hours/56 days (rat) 1-CHLORO-1,1-DIFLUOROETHANE (HCFC 142B) (75-68-3) LC50 (Inhalation): 1758000 mg/m3/2 hours (mouse)

LCLo (Inhalation): 212000 mg/m3/30M (rat)

TCLo (Inhalation): 1000 ppm/6 hours/3-15 days of pregnancy (rat)

12. ECOLOGICAL INFORMATION

Environment

Dangerous for the ozone layer. Hydrogenated chlorofluorocarbon compounds (HCFC's) do not persist in the stratosphere to the same degree as chlorofluorocarbons (CFC's). Although ozone depleting, they have a lower ozone depleting effect than CFC's. Release of HCFCs into the environment should be minimised and where possible, recycling of HCFCs is recommended.

13. DISPOSAL CONSIDERATIONS

Waste Disposal

It is an offence under Federal law to knowingly exhaust this product to atmosphere whether by intent or negligence. Recover and recycle using appropriate techniques and equipment. Notify the manufacturer that you will be returning a faulty cylinder. Residual product will be disposed of when the cylinder is returned. It is an offence under federal law to knowingly exhaust this product to atmosphere whether by intent or negligence, for example, poorly maintained, leaking equipment.

Legislation Dispose of in accordance with relevant local legislation.



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14. TRANSPORT INFORMATION

Transport

Ensure cylinder is separated from driver and that outlet of relief device is not obstructed.



CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE

Shipping Name REFRIGERANT GAS, N.O.S.

UN No. 1078 **DG Class** 2.2 Subsidiary Risk(s) None Allocated

Packing Group None Allocated **Hazchem Code** 2RE **EPG** 2C2

15. REGULATORY INFORMATION

Poison Schedule A poison schedule number has not been allocated to this product using the criteria in the Standard for the Uniform

Scheduling of Drugs and Poisons (SUSDP).

AICS All chemicals listed on the Australian Inventory of Chemical Substances (AICS).

16. OTHER INFORMATION

Additional Information The storage of significant quantities of gas cylinders must comply with AS4332 The storage and handling of gases in cylinders. This product is a replacement for R500.

APPLICATION METHOD: Transferred as a liquid into and out of refrigeration equipment by controlled pressure decanting through flexible pipework.

ABBREVIATIONS:

ADB - Air-Dry Basis.

BEI - Biological Exposure Indice(s)

CAS# - Chemical Abstract Service number - used to uniquely identify chemical compounds.

CNS - Central Nervous System.

EINECS - European INventory of Existing Commercial chemical Substances.

IARC - International Agency for Research on Cancer.

M - moles per litre, a unit of concentration.

mg/m3 - Milligrams per cubic metre.

NOS - Not Otherwise Specified.

NTP - National Toxicology Program.

OSHA - Occupational Safety and Health Administration.

pH - relates to hydrogen ion concentration using a scale of 0 (high acidic) to 14 (highly alkaline).

ppm - Parts Per Million.

RTECS - Registry of Toxic Effects of Chemical Substances.

TWA/ES - Time Weighted Average or Exposure Standard.

HEALTH EFFECTS FROM EXPOSURE:

It should be noted that the effects from exposure to this product will depend on several factors including: frequency and duration of use; quantity used; effectiveness of control measures; protective equipment used and method of application. Given that it is impractical to prepare a Chem Alert report which would encompass all possible scenarios, it is anticipated that users will assess the risks and apply control methods where appropriate.

PERSONAL PROTECTIVE EQUIPMENT GUIDELINES:

The recommendation for protective equipment contained within this Chem Alert report is provided as a guide only. Factors such as method of application, working environment, quantity used, product concentration and the availability of engineering controls should be considered before final selection of personal protective equipment is

Report Status

This document has been compiled by RMT on behalf of the manufacturer of the product and serves as the manufacturer's Safety Data Sheet ('SDS').

It is based on information concerning the product which has been provided to RMT by the manufacturer or obtained from third party sources and is believed to represent the current state of knowledge as to the appropriate safety and handling precautions for the product at the time of issue. Further clarification regarding any aspect of the product should be obtained directly from the manufacturer.



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While RMT has taken all due care to include accurate and up-to-date information in this SDS, it does not provide any warranty as to accuracy or completeness. As far as lawfully possible, RMT accepts no liability for any loss, injury or damage (including consequential loss) which may be suffered or incurred by any person as a consequence of their reliance on the information contained in this SDS.

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> SDS Date: 26 Mar 2010 End of Report

