

SAFETY DATA SHEET

114

MEDICAL NITROUS OXIDE, COMPRESSED **Product Name**

1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

Supplier Name BOC LIMITED (AUSTRALIA)

Address 10 Julius Avenue, North Ryde, NSW, AUSTRALIA, 2113

Telephone 131 262, (02) 8874 4400 Fax 132 427 (24 hours)

Emergency 1800 653 572 (24/7) (Australia only)

Web Site http://www.boc.com.au/

114 - SDS NUMBER • NITROUS OXIDE (FORMERLY) • PRODUCT CODE: 610 Synonym(s)

ANAESTHETIC • MEDICAL APPLICATIONS Use(s)

SDS Date 26 Mar 2010

2. HAZARDS IDENTIFICATION

CLASSIFIED AS HAZARDOUS ACCORDING TO ASCC CRITERIA

RISK PHRASES

R63 Possible risk of harm to the unborn child.

R8 Contact with combustible material may cause fire.

SAFETY PHRASES

S17 Keep away from combustible material. S51 Use only in well ventilated areas. 59 Keep container in a well ventilated place.

CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE

UN No. 1070 **DG Class** 2.2 Subsidiary Risk(s) 5.1 Hazchem Code 2P **EPG** 2C8 **Packing Group** None Allocated

3. COMPOSITION/INFORMATION ON INGREDIENTS

Ingredient	Formula	CAS No.	Content
NITROUS OXIDE	N20	10024-97-2	>99.5%

4. FIRST AID MEASURES

Eye Cold burns: Immediately flush with tepid water or with sterile saline solution. Hold eyelids apart and irrigate for 15

minutes. Seek medical attention.

If inhaled, remove from contaminated area. To protect rescuer, use an Air-line respirator or Self Contained Inhalation

Breathing Apparatus (SCBA). Be aware of possible explosive atmospheres. Apply artificial respiration if not breathing. Give oxygen if available. For advice, contact a Poisons Information Centre on 13 11 26 (Australia Wide)

or a doctor.

Skin Cold burns: Remove contaminated clothing and gently flush affected areas with warm water (30°C) for 15 minutes.

Apply sterile dressing and treat as for a thermal burn. For large burns, immerse in warm water for 15 minutes. DO

NOT apply any form of direct heat. Seek immediate medical attention.

Ingestion Due to product form and application, ingestion is considered unlikely.

ChemAlert.

Page 1 of 4 **RMT**

Reviewed: 26 Mar 2010

Printed: 26 Mar 2010

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Advice to Doctor Treat symptomatically

5. FIRE FIGHTING MEASURES

Flammability Non flammable - oxidising agent. Supports combustion and may cause fire/explosion in contact with incompatible

substances, strong acids, reducing agents, combustibles and flammables.

Fire and Explosion

Temperatures in a fire may cause cylinders to rupture. Cool cylinders or containers exposed to fire by applying

water from a protected location. Do not approach cylinders or containers suspected of being hot.

Extinguishing Use water fog to cool containers from protected area.

Hazchem Code 2P

6. ACCIDENTAL RELEASE MEASURES

Spillage

If the cylinder is leaking, evacuate area of personnel. Inform manufacturer/supplier of leak. Use personal protective equipment. Carefully move material to a well ventilated remote area, then allow to discharge. Do not attempt to repair leaking valve or cylinder safety devices.

7. STORAGE AND HANDLING

Storage Do not store near sources of ignition or incompatible materials. Cylinders should be stored below 45°C in a secure

area, upright and restrained to prevent cylinders from falling. Cylinders should also be stored in a dry, well ventilated area constructed of non-combustible material with firm level floor (preferably concrete), away from areas

of heavy traffic and emergency exits.

Handling Use of safe work practices are recommended to avoid eye or skin contact and inhalation. Do not drag, drop, slide

or roll cylinders. The uncontrolled release of a gas under pressure may cause physical harm. Use a suitable hand

truck for cylinder movement.

8. EXPOSURE CONTROLS/ PERSONAL PROTECTION

Exposure Stds

Ingradiant	Reference	TWA		STEL	
Ingredient	Reference	ppm	mg/m3	ppm	mg/m3
Nitrous oxide	ASCC (AUS)	25	45		

Biological Limits No biological limit allocated.

Engineering Controls

Avoid inhalation. Use in well ventilated areas. Where an inhalation risk exists, mechanical extraction ventilation is recommended. Maintain vapour levels below the recommended exposure standard.

PPE

Wear safety boots, insulated or leather gloves and safety glasses. Where an inhalation risk exists, wear: a Type NO (Nitrogen Oxides) respirator. At high vapour levels, wear: an Air-line respirator or self Contained Breathing Apparatus (SCBA).







9. PHYSICAL AND CHEMICAL PROPERTIES

Appearance	COLOURLESS GAS	Solubility (Water)	0.59 L/L
Odour	SLIGHT SWEET ODOUR	Specific Gravity	NOT APPLICABLE
рН	NOT APPLICABLE	% Volatiles	100 %
Vapour Pressure	5700 kPa @ 25°C	Flammability	NON FLAMMABLE
Vapour Density	NOT AVAILABLE	Flash Point	NOT RELEVANT
Boiling Point	-88.5°C	Upper Explosion Limit	NOT RELEVANT
Melting Point	NOT AVAILABLE	Lower Explosion Limit	NOT RELEVANT
Evaporation Rate	NOT APPLICABLE		
Critical Pressure	7254 kPa	Critical Temperature	36.4°C

Density 1.53 (Air = 1)



Product Name MEDICAL NITROUS OXIDE, COMPRESSED

10. STABILITY AND REACTIVITY

Chemical Stability Stable under recommended conditions of storage.

Conditions to Avoid Avoid heat, sparks, open flames and other ignition sources.

Material to Avoid Incompatible with with powerful reducing agents such as phosphine, stannous chloride and hydrogen.

Ensure that all materials/lubricants in contact with this product are oxygen compatible. Rust and scale

may cause ignitions.

Decomposition May evolve toxic gases if heated to decomposition.

Hazardous Reactions Polymerization will not occur.

11. TOXICOLOGICAL INFORMATION

Health Hazard Summary Asphyxiant gas - anaesthetic. Nitrous oxide passes into all gas containing spaces in the body faster than nitrogen passes out, thus it should not be used with any condition where its expansion might be dangerous. Epidemiological studies suggest an increased risk of spontaneous abortion and low birth weight in off-spring in female workers employed in operating theatres and dental surgeries. These findings are controversial.

Eye Non irritant. However, direct contact with evaporating liquid may result in severe cold burns with possible

permanent damage.

Inhalation Non irritant - anaesthetic. After prolonged use nitrous oxide may inactivate vitamin B12, resulting in megaloblastic

changes in bone marrow. Exceptionally heavy occupational exposure and addiction may result in myeloneuropathy and sub-acute combined degeneration. This would require levels in excess of 5,000ppm to

frequent (more than once every two days) exposure to analgesic concentrations.

Skin Non irritant. However, direct contact with the liquefied material or escaping compressed gas may cause frostbite

injury.

Ingestion Ingestion is considered unlikely due to product form.

Toxicity Data NITROUS OXIDE (10024-97-2)

LC50 (Inhalation): 1068 mg/m3 (rat)

TCLo (Inhalation): 1 pph/8 hours (rat - reproductive effects)

12. ECOLOGICAL INFORMATION

Environment

ATMOSPHERE: Nitrous oxide is toxic and is considered an air pollutant. Fairly stable in the atmosphere. May be toxic to terrestrial animals. SOIL: Due to its very low boiling point it is expected to quickly evaporate if released on soil. WATER: May dissolve in water, although evaporation will be a major removal factor.

13. DISPOSAL CONSIDERATIONS

Waste Disposal Cylinders should be returned to the manufacturer or supplier for disposal of contents.

Legislation Dispose of in accordance with relevant local legislation.

14. TRANSPORT INFORMATION

Transport

Ensure cylinder is separated from driver and that outlet of relief device is not obstructed. Refer to Commonwealth, State and Territory Dangerous Goods Legislation which contain requirements which affect gas storage and transport.





CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE

Shipping Name NITROUS OXIDE

UN No. 1070 DG Class 2.2 Subsidiary Risk(s) 5.1 Packing Group None Allocated Hazchem Code 2P EPG 2C8

15. REGULATORY INFORMATION

Poison Schedule Classified as a Schedule 4 (S4) Poison using the criteria in the Standard for the Uniform Scheduling of Drugs and

Poisons (SUSDP).

AICS All chemicals listed on the Australian Inventory of Chemical Substances (AICS).

Page 3 of 4

RMT

Reviewed: 26 Mar 2010 Printed: 26 Mar 2010



MEDICAL NITROUS OXIDE, COMPRESSED

16. OTHER INFORMATION

Additional Information

The storage of significant quantities of gas cylinders must comply with AS4332 The storage and handling of gases in cylinders.

APPLICATION METHOD: Gas regulator and demand flow control to mask or mouthpiece.

ABBREVIATIONS:

ADB - Air-Dry Basis.

BEI - Biological Exposure Indice(s)

CAS# - Chemical Abstract Service number - used to uniquely identify chemical compounds.

CNS - Central Nervous System.

EINECS - European INventory of Existing Commercial chemical Substances.

IARC - International Agency for Research on Cancer.

M - moles per litre, a unit of concentration.

mg/m3 - Milligrams per cubic metre.

NOS - Not Otherwise Specified.

NTP - National Toxicology Program.

OSHA - Occupational Safety and Health Administration.

pH - relates to hydrogen ion concentration using a scale of 0 (high acidic) to 14 (highly alkaline).

ppm - Parts Per Million.

RTECS - Registry of Toxic Effects of Chemical Substances.

TWA/ES - Time Weighted Average or Exposure Standard.

HEALTH EFFECTS FROM EXPOSURE:

It should be noted that the effects from exposure to this product will depend on several factors including: frequency and duration of use; quantity used; effectiveness of control measures; protective equipment used and method of application. Given that it is impractical to prepare a Chem Alert report which would encompass all possible scenarios, it is anticipated that users will assess the risks and apply control methods where appropriate.

PERSONAL PROTECTIVE EQUIPMENT GUIDELINES:

The recommendation for protective equipment contained within this Chem Alert report is provided as a guide only. Factors such as method of application, working environment, quantity used, product concentration and the availability of engineering controls should be considered before final selection of personal protective equipment is

Report Status

This document has been compiled by RMT on behalf of the manufacturer of the product and serves as the manufacturer's Safety Data Sheet ('SDS').

It is based on information concerning the product which has been provided to RMT by the manufacturer or obtained from third party sources and is believed to represent the current state of knowledge as to the appropriate safety and handling precautions for the product at the time of issue. Further clarification regarding any aspect of the product should be obtained directly from the manufacturer.

While RMT has taken all due care to include accurate and up-to-date information in this SDS, it does not provide any warranty as to accuracy or completeness. As far as lawfully possible, RMT accepts no liability for any loss, injury or damage (including consequential loss) which may be suffered or incurred by any person as a consequence of their reliance on the information contained in this SDS.

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SDS Date: 26 Mar 2010 **End of Report**



Page 4 of 4

Reviewed: 26 Mar 2010

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